

PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>5</sup> : <b>C07J 71/00, A61K 31/58</b>		A1	(11) International Publication Number: <b>WO 94/15947</b> (43) International Publication Date: <b>21 July 1994 (21.07.94)</b>
(21) International Application Number: <b>PCT/SE93/01081</b>		(81) Designated States: AT, AU, BB, BG, BR, BY, CA, CH, CZ, DE, DK, ES, FI, GB, HU, JP, KP, KR, KZ, LK, LU, LV, MG, MN, MW, NL, NO, NZ, PL, PT, RO, RU, SD, SE, SK, UA, UZ, VN, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).	
(22) International Filing Date: <b>17 December 1993 (17.12.93)</b>		Published <i>With international search report.</i>	
(30) Priority Data: 9300030-5 8 January 1993 (08.01.93) SE 9300082-6 14 January 1993 (14.01.93) SE		(71) Applicant: <b>AKTIEBOLAGET ASTRA [SE/SE]; S-151 85 Södertälje (SE).</b>	
(72) Inventors: <b>BRATTSAND, Ralph, Lennart; Skolmästarvägen 4, S-224 67 Lund (SE). EDMAN, Peter; Kamrersvägen 18, S-237 00 Bjärred (SE). HÖGBERG, Thomas; Byvägen 11, S-232 51 Åkarp (SE). NILSSON, Stinabritt; Utställaregränden 9, S-226 47 Lund (SE). THALEN, Bror, Arne; Morkullevägen 35, S-237 00 Bjärred (SE). ULMIUS, Jan, Erik; Planvägen 6, S-222 47 Lund (SE).</b>			
(54) Title: <b>NOVEL COLON- OR ILEUM-SPECIFIC STEROID DERIVATIVES</b>			
(57) Abstract <p>Novel compounds which are a glucocorticosteroid (GCS) chemically bound to a sugar, having the general formula: GCS<sup>1</sup>-O-Sugar<sup>1</sup> for colon- or ileum-specific delivery of the GCS to inflamed bowel mucosa, as well as processes for their preparation, pharmaceutical preparations containing the compounds and the use of said compounds in therapy.</p>			

*FOR THE PURPOSES OF INFORMATION ONLY*

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgyzstan	RU	Russian Federation
CF	Central African Republic	NP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovakia
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechoslovakia	LV	Latvia	TG	Togo
CZ	Czech Republic	MC	Monaco	TJ	Tajikistan
DE	Germany	MD	Republic of Moldova	TT	Trinidad and Tobago
DK	Denmark	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	US	United States of America
FI	Finland	MN	Mongolia	UZ	Uzbekistan
FR	France			VN	Viet Nam
GA	Gabon				

Novel colon- or ileum-specific steroid derivatives5    Field of the invention

The present invention relates to novel compounds which are a glucocorticosteroid (GCS) chemically bound to a sugar, for local colon- or ileum-specific delivery of the GCS to inflamed bowel mucosa and to processes for their preparation. The invention also relates to pharmaceutical preparations containing the compounds, and to the use of said compounds in therapy. Also pharmaceutically and pharmacologically acceptable salts of the compounds according to the invention are comprised.

The object of the invention is to provide an anti-inflammatory GCS, with a high first pass metabolism in the liver, chemically bound to a sugar, or a pharmaceutical composition of the GCS-sugar compound for local colon- or ileum-specific delivery of the GCS to the inflamed bowel mucosa.

25    Background of the invention

Ulcerative colitis (UC) is a serious inflammatory disease affecting the colon and then most often the descendens and sigmoideum segments of colon. Morbus Crohn is a dangerous inflammatory bowel disease, sometimes affecting primarily colon but most often affecting the terminal small bowel - the ileum. These inflammatory processes are sensitive to GCS therapy, but hitherto effective long term treatment has been hampered by serious adverse GCS effects in the systemic circulation (eg osteoporosis, precipitation of diabetes, blocked HPA-axis etc).

In order to locally treat the mainly affected distal part of colon, the luminal concentration of steroid in colon must be high enough to allow for intraluminal transport despite a competing systemic absorption in the colon

5 ascendens. The ideal profile for colon-specific therapy would be reached by release of a potent GCS with a very high first pass metabolic inactivation in the liver. There should be a continuous and complete release of the active GCS during the colon passage. The best therapy has

10 hitherto been attained with budesonide, which has a favourable combination of high topical potency and substantial hepatic first pass inactivation, Can J Gastroenterol 4:407-414, 1990. To reach colon mucosa of the distal segments by local therapy, budesonide has to

15 be encapsulated in a pharmaceutical formulation, which when given orally starts to release budesonide in the terminal ileum. Such a pharmaceutical formulation is disclosed in PCT/SE90/00738. However, with a pharmaceutical formulation of that kind it is difficult

20 to get a complete GCS release during the colonic transit, which at least in periods of active disease is short and quite variable. Thus, a substantial fraction of GCS is often bypassing the patient without being released.

25 An approach to more specific therapy of colon, has been a chemical targeting based on bacteria-specific cleavage of a GCS prodrug, e.g. a  $\beta$ -D-glucoside. In EP 123485 and also in J Med Chem 28:51-57, 1985, in Pharmaceutical Res 8:445-454, 1991, and in Advanced Drug Delivery Reviews 7:149-

30 199, 1991, such prodrugs have recently been described based on dexamethasone and hydrocortisone. However, these GCS-glycosides will not be colon-specific as stated, because the released glucocorticosteroids have too low first pass inactivation in the liver (Can. J. Gastroenterol. 4:407-414, 1990). In man a substantial fraction of the GCS released can be anticipated to reach the systemic circulation intact and by that provoke

adverse reactions. Furthermore, plain delivery of GCS-glycoside will not lead to the right type of continuous colonic release. When the glycoside meets glycosidase-containing bacteria in cecum and ascending colon, a rapid 5 intraluminal hydrolysis and GCS absorption will occur. This reduces markedly the possibilities for subsequent local release in colon transversum, descendens, sigmoideum and rectum, which parts are all more prone to colitis than what ascendens is. This poor local spreading 10 of active GCS from glycoside prodrug has not been discussed earlier.

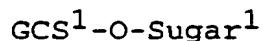
The most common location of lesions in Morbus Crohn is the ileum. Once the ileum is affected, these patients are 15 very often operated by resection of terminal ileum including the ileo-cekal valve, which is the valve normally blocking colonic bacterial backwash into the ileum. There is a recent piece of information that this fecal contamination into bowel segments not normally exposed to high bacterial counts, contributes to the 20 common retrograde spreading of serious inflammation and recurrence of clinical disease. Often these patients have to be operated by further ileal resection or to widening of ileal lumen. Current GCS treatment of Morbus Crohn of 25 small bowel is based on conventional tablets releasing their steroid content in upper bowel segments. Because these tablets work via the systemic route and high doses have to be given, serious adverse effects are provoked. Recently retarded formulations have been tested for 30 improving direct release to ileal mucosa. However, with the current type of retarded formulations controlled by pH and osmotic forces, it is not possible to reach a concentrated release of active GCS at the front of bacterial invasion of small bowel. The use of steroid 35 glycosides in local treatment of ileal Morbus Crohn has not been discussed earlier.

Disclosure of the invention

According to the present invention new compounds are disclosed providing a new way to reach a colon-specific  
5 delivery better related to the appropriate distribution of mucosal inflammation.

The ideal profile for local treatment of small bowel inflammation in Morbus Crohn (especially in resected  
10 patients or in patients with poor function of the ileo-cekal valve) is a GCS-glycoside releasing a potent GCS with very high first pass metabolism in the liver. When a compound of that kind meets the bacterial front at ileal level, it is anticipated that much higher local  
15 concentrations of active GCS can be reached at the bacterial front than by earlier types of pharmaceutical formulations.

The compounds according to the invention have the general  
20 formula

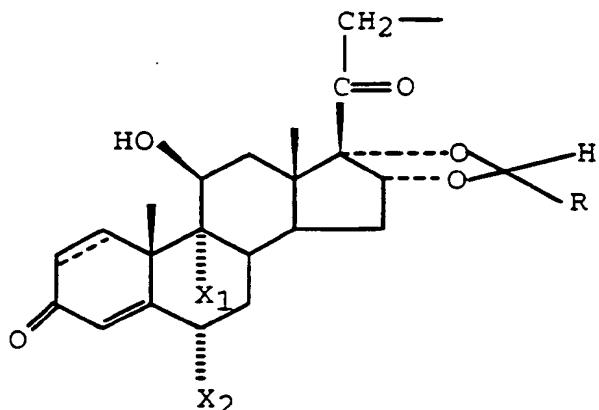


where GCS<sup>1</sup> is a steroid (GCS<sup>1</sup>-OH) with a high first-pass  
25 metabolism in the liver and Sugar<sup>1</sup> is recognizable as substrate by bacterial glycosidases and linked to the 21-position of the steroid via a glycosidic bond that is hydrolyzed by glycosidases in the colonic microflora.

30 GCS<sup>1</sup> can be chosen as a steroid with a 16,17-acetal grouping, providing an additional easily metabolized moiety, which is selected from the group of formula I

5

5



10

15

20

25

R being a hydrocarbon chain with 1 to 9 carbon atoms, the C<sub>1</sub>-C<sub>2</sub> bond being a single or a double bond, X<sub>1</sub> and X<sub>2</sub> being the same or different hydrogen, fluoro, chloro or bromo substituents and X<sub>3</sub> being a fluoro, chloro or bromo substituent.

The 1,2-position of the GCS<sup>1</sup> is saturated or is a double bond.

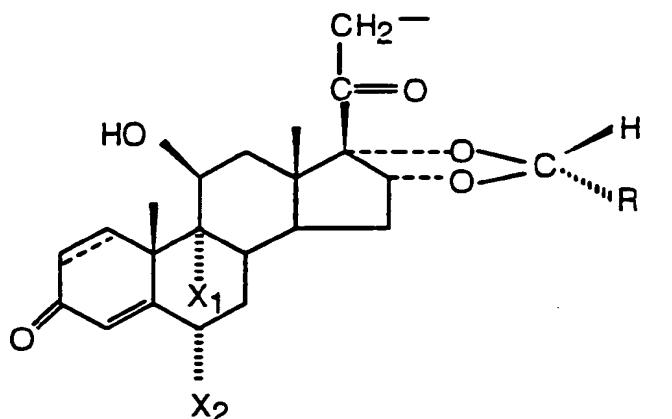
35

The acetal I is in epimerically pure form i.e. the acetal I is the corresponding pure 22R-epimer, IA, or 22S-epimer, IB

5

10

15

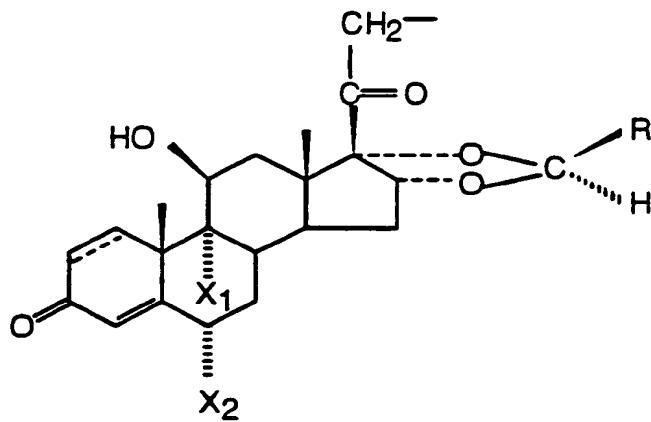


IA

20

25

30



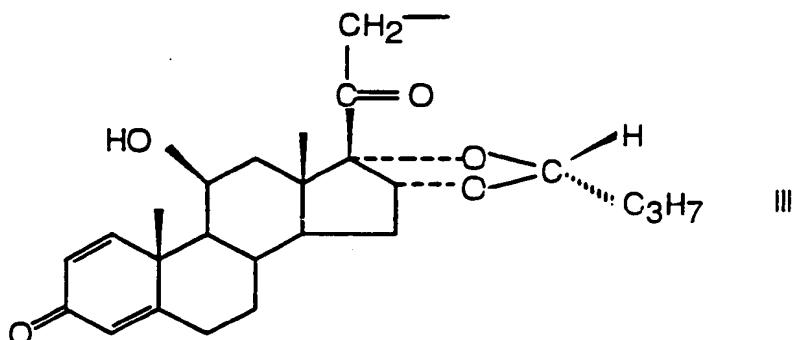
IB

or is in the form of an epimeric mixture.

Preferably the acetal I is the 22R-epimer.

Most preferred GCS of the invention is the 22R-epimer of budesonide (GCS<sup>1</sup>-OH) with the formula III

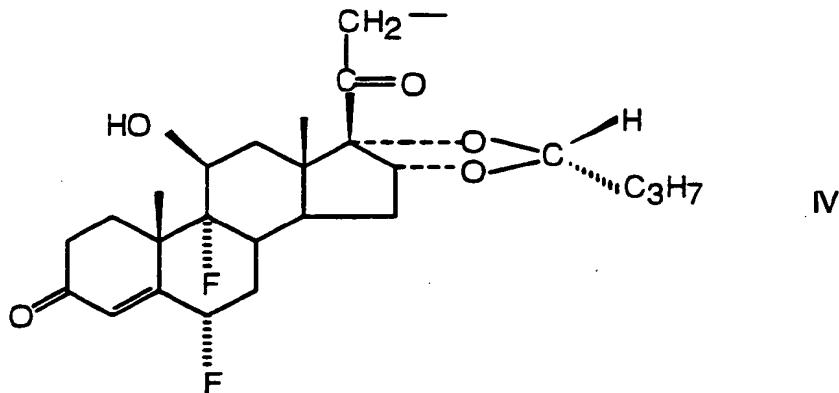
5



10

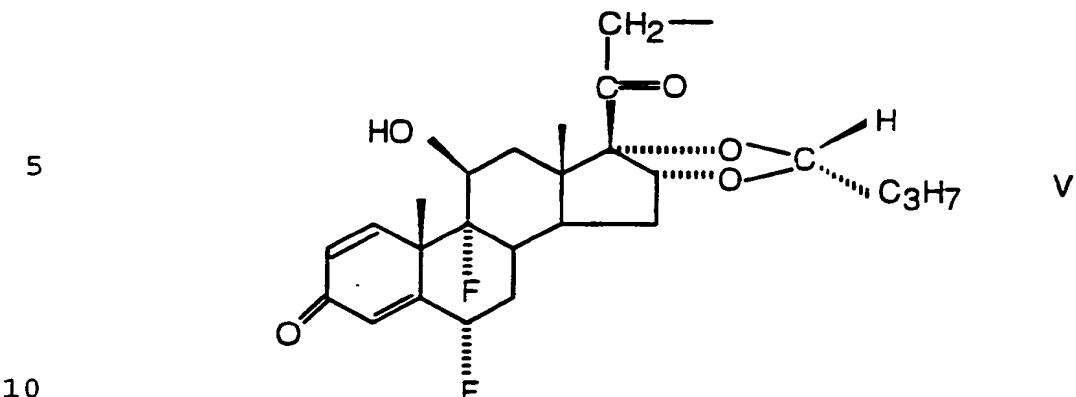
or the 22R-epimer of  $16\alpha,17\alpha$ -butylidenedioxy- $6\alpha,9\alpha$ -difluoro- $11\beta$ -hydroxy-4-pregnene-3,20-dione-21-yl, hereinafter called the 22R-epimer of GCS<sup>1</sup> IV, with the  
15 formula

20



25

or the 22R-epimer of  $16\alpha,17\alpha$ -butylidenedioxy- $6\alpha,9\alpha$ -difluoro- $11\beta$ -hydroxy-1,4-pregnadiene-3,20-dione-21-yl, hereinafter called the 22R-epimer of GCS<sup>1</sup> V, with the  
30 formula



Sugar<sup>1</sup>-OH can be chosen as a monosaccharide, a disaccharide or an oligosaccharide, e.g D-glucose, D-glucuronic acid,  
15 D-galactose, D-galacturonic acid, D-cellobiose or D-lactose.

Preferably the Sugar<sup>1</sup> is  $\beta$ -linked D-glucose or D-glucuronic acid.

20 Most preferred compounds according to the invention are budesonide 22R-epimer  $\beta$ -D-glucoside, GCS<sup>1</sup> IV 22R-epimer  $\beta$ -D-glucoside and GCS<sup>1</sup> V 22R-epimer  $\beta$ -D-glucoside, budesonide 22R-epimer  $\beta$ -D-glucosiduronic acid, GCS<sup>1</sup> IV  
25 22R-epimer  $\beta$ -D-glucosiduronic acid and GCS<sup>1</sup> V 22R-epimer  $\beta$ -D-glucosiduronic acid.

The compounds according to the present invention include an active GCS, which when released possesses high topical  
30 anti-inflammatory potency as well as undergoes profound hepatic first pass inactivation (85% or more). The combination of the GCS with a substantial first pass metabolism and a colon directed delivery provided by a bacteria specific enzymatic cleavage of the compound  
35 makes this possible.

Also comprised according to the present invention are pharmacologically and pharmaceutically acceptable salts of the compounds having the general formula GCS<sup>1</sup>-O-Sugar<sup>1</sup>.

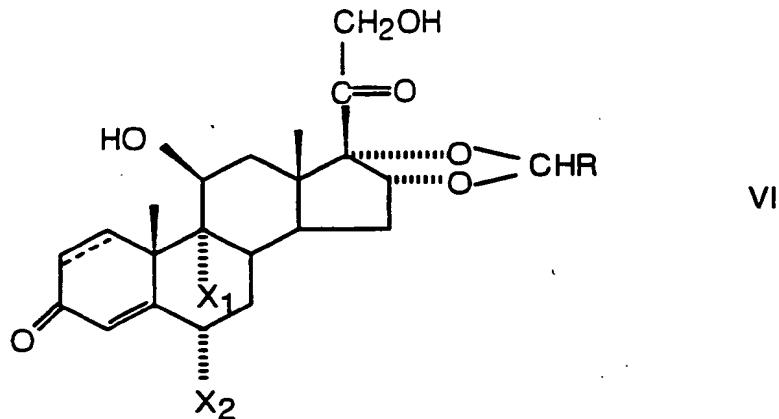
5

Methods of preparation

The compounds according to the invention are prepared by the condensation of a mono-, di- or oligosaccharide with 10 a compound of the formulas VI, VI A, VI B and VII

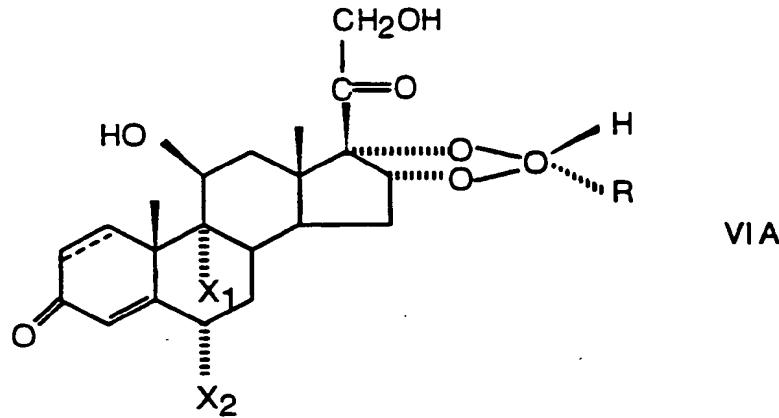
15

20



25

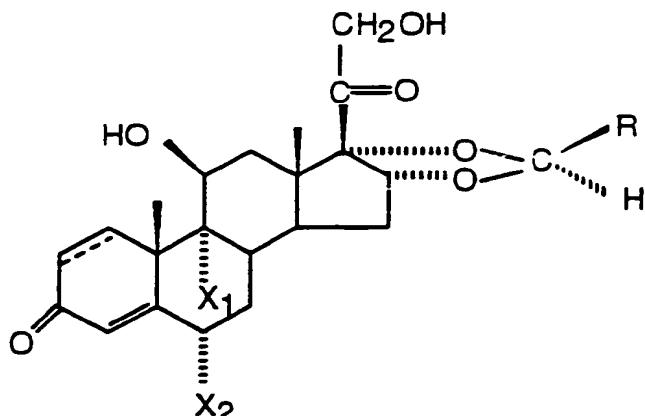
30



35

10

5

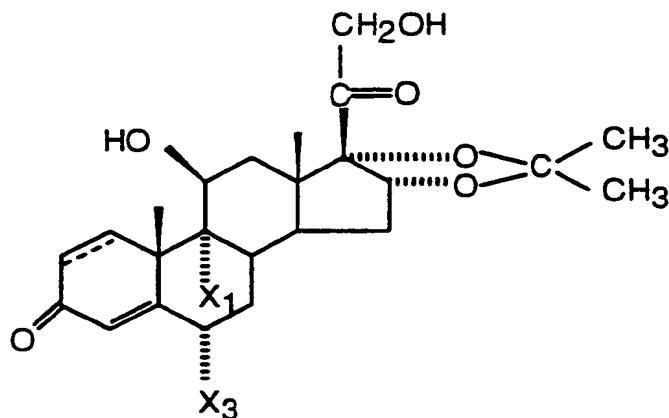


VI B

10

15

20



VII

25

wherein the solid and broken lines between carbon 1 and carbon 2 represent a single or a double bond. R,  $X_1$ ,  $X_2$  and  $X_3$  are as defined above.

30 The process according to the present invention to convert a compound of formulas VI, VI A, VI B and VII to the corresponding 21-glycosides is carried out by the condensation of a suitably protected derivative of the mono-, di- or oligosaccharide with the steroid or a derivative of the steroid, followed by deprotection of the condensation product.

35

Most suitable are glycosidation methods where the anomeric hydroxyl group of the glycosyl donor is exchanged for a better leaving group or a group which is transformed into a leaving group under the influence of a promoter. Preferably, glycosyl bromides and chlorides are condensed with alcohols with promoters such as silver trifluoromethanesulfonate, silver perchlorate, silver carbonate, mercury(II)bromide/mercury(II)cyanide, silver zeolite, zinc chloride or tetraethyl ammonium bromide.

Glycosyl esters react with alcohols preferably under promotion of Lewis acids e.g. trimethylsilyl trifluoromethanesulfonate, tin(IV)chloride, tin(IV)chloride/silver perchlorate or boron trifluoride etherate. Alkyl and aryl thioglycosides can be reacted with alcohols using various thiophilic promoters, preferably N-iodosuccinimide/trifluoromethanesulfonic acid, iodonium dicollidine perchlorate, methylsulfenyl trifluoromethanesulfonate, methylsulfenyl bromide, benzeneselenyl trifluoromethanesulfonate, nitrosyl tetrafluoroborate, methyl trifluoromethanesulfonate, sulfuryl chloride/trifluoromethanesulfonic acid, dimethyl(methylthio)sulfonium trifluoromethanesulfonate or dimethyl(methylthio)sulfonium tetrafluoroborate. Glycosyl fluorides can use preferably trimethylsilyl trifluoromethanesulfonate, boron trifluoride etherate, tetrafluorosilane, titan tetrafluoride, trifluoromethanesulfonic anhydride, tin(II)chloride/silver trifluoromethanesulfonate or tin(II)chloride/silver perchlorate promotion. Glycosyl trichloroacetimidates can use Lewis acids such as trimethylsilyl trifluoromethanesulfonate or boron trifluoride etherate. n-Pentenyl glycosides can be activated with halonium ions preferably N-bromosuccinimide, iodonium dicollidine perchlorate or N-iodosuccinimide combined with trifluoromethanesulfonic acid, silver trifluoromethanesulfonate or triethylsilyl trifluoromethanesulfonate. Furthermore 1,2-orthoesters,

1,2-oxazolines, 1,2-thioorthoesters, 1,2-cyanoethylidene derivatives, glycosyl thiocyanates, glycosyl sulfoxides, glycosyl sulfones, S-glycosyl xanthates, S-glycosyl dithiocarbamates, anhydrosugars and glycals can be used  
5 as glycosyl donors.

The pattern of protective groups of the glycosyl donor is of importance for the stereoselectivity of the glycosidic bond. Especially important is the protective group at the  
10 2-position of the glycosyl donor. For example an acetyl or a benzoyl group at the 2-position of e.g. a glucosyl, glucosyluronate, galactosyl, galactosyluronate, cellobiosyl or lactosyl donor gives predominantly  $\beta$ -condensation. By using a so-called non-participating  
15 group e.g. allyl or benzyl at the 2-position of e.g. a galactosyl, galactosyluronate, glucosyl, glucosyluronate, cellobiosyl or lactosyl donor, these can be coupled mainly  $\alpha$  to the steroid molecule. The solvent used for  
the condensation reaction is an aprotic solvent,  
20 preferably dichloromethane, chloroform, carbon tetrachloride, N,N-dimethylformamide, nitromethane, ethyl acetate, tetrahydrofuran, diethyl ether, toluene, dioxane, 1,2-dichloroethane, acetonitrile, monoglyme or a mixture of these. The solvent and the temperature often  
25 influence the stereochemical outcome of the reaction. For example, in the case of a galactosyl donor with a non-participating group at the 2-position e.g. diethyl ether often promotes  $\alpha$ -condensation, whereas e.g. acetonitrile often promotes  $\beta$ -condensation.

30

In an alternative glycosidation method, the anomeric hydroxyl group of the glycosyl donor is reacted with a base e.g. sodium hydride and a derivative of the steroid where the 21-position has a suitable leaving group e.g. a  
35 trifluoromethanesulfonyl group. The glycosyl donor with an anomeric hydroxyl group can also be coupled to the steroid using various condensating reagents e.g.

triphenylphosphine and diethyl azodicarboxylate. The mono-, di- or oligosaccharide can also be condensed with the steroid with a catalytic amount of e.g. trifluoromethanesulfonic acid in a suitable solvent e.g. 5 dimethyl sulfoxide.

The protective groups of the condensation product can be removed by known methods. For example acyl protective groups are suitably removed by transesterification with 10 e.g. sodium methoxide.

#### Pharmaceutical preparations

Further according to the invention conventional pharmaceutical preparations or pharmaceutical 15 preparations that modestly retard the initial release of prodrug in cecum and colon ascendens, so that there will be a much more complete and continuous exposition of active GCS over the most important colonic and sigmoidal 20 regions are disclosed for the proper treatment of colonic inflammation

This is accomplished by a pharmaceutical preparation containing the GCS prodrug protected with a coating that 25 bursts after a pre-determined time, i.e. 5-10 hours after the preparation has left the stomach, when the preparation resides in the colon ascendens. The preparation is protected in the stomach by an enteric coating.

30 The objective is also accomplished by a pharmaceutical preparation containing the GCS prodrug protected by a polysaccharide that can be degraded by the gut microflora. The degree of protection should be adjusted 35 so that the main part of release occurs after colon ascendens. The preparation could optionally be protected by an enteric coating.

The pharmaceutical preparations according to present invention are described more detailed in the following:

a) The GCS prodrug is formulated in a core through the  
5 well-known techniques granulation or granulation + extrusion + marumerization with suitable excipients including a super disintegrand e.g. crosslinked polyvinylpyrrolidone, sodium-CMC or sodium starch glycolate. The core is coated with a layer that will  
10 control the water penetration rate into the core. The layer can consist of an insoluble polymer e.g. ethylcellulose, hydroxypropylcellulose, Eudragit RS or Eudragit RL together with a hydrophobic agent e.g. a metal stearate. The proportions of the polymer and the  
15 metal stearate and/or the thickness of the layer will determine the lag time until the water has penetrated the layer and entered the core where the disintegrating agent will swell and rupture the membrane, releasing the GCS prodrug. The core and the layer is also coated with an enteric polymer e.g. Eudragit L, Eudragit S, cellulose acetate phtalate or hydroxypropylmethylcellulose phtalate which will prevent the water penetration when the  
20 formulation resides in the stomach.

b) The GCS prodrug is layered on a suitable core together with a suitable binding agent e.g. PVP or a water soluble cellulose ether in a fluid bed process or a rotor process. This core is coated with a layer containing a gut microflora degradable polysaccharide e.g. pectin,  
25 guar gum, dextran, carrageenan, amylose or chitosan in an insoluble polymer e.g. ethylcellulose, Eudragit R, Eudragit S or Eudragit NE. The time for degradation of the polysaccharide, so that the GCS prodrug can be released, can be altered by the proportion of the  
30 polysaccharide and insoluble polymer and/or the thickness of the layer. Optionally the layer can be protected by a layer of an enteric polymer e.g. Eudragit L, Eudragit S,

cellulose acetate phtalate or hydroxypropylmethylcellulose phtalate.

Working examples

5

The invention will be further illustrated by the following non-limitative examples. Concentrations were performed under reduced pressure at <40°C bath temperature. Melting points were obtained with a Mettler 10 FP82 Olympus BH-2 hot stage microscope. NMR spectra were recorded with a Varian VXR-300 instrument. The following reference signals were used:

Me<sub>4</sub>Si, δ 0.00 (<sup>1</sup>H in CDCl<sub>3</sub>); and MeOH, δ 3.35 (<sup>1</sup>H in CD<sub>3</sub>OD). In the assignments below, atoms of glucose and glucuronic acid carry the ' superscript. Molecular weights were determined by fast atom bombardment (FAB) spectrometry. Column chromatography was performed on silica gel (60Å, 40-63 µm; Merck, Darmstadt, Germany). HPLC-analyses were performed on a C<sub>18</sub> column (µBondapak 10 µm 150 x 3.9 mm or Supelcosil 5 µm 150x4.6 mm) using acetonitrile/water or acetonitrile/20 mM TBAHS + 10 mM phosphate buffer pH 7 as eluent. Powdered molecular sieves (4Å; Fluka, Buchs, Switzerland) were heated to 25 300°C under vacuum overnight. Dichloromethane and toluene were dried over 4Å molecular sieves, and methanol over 3Å molecular sieves.

Example 1.

30 (22R)-16α,17α-Butylidenedioxy-6α,9α-difluoro-11β-hydroxy-4-pregnene-3,20-dione-21-yl β-D-glucopyranoside. (GCS<sup>1</sup> IV 22R-epimer β-D-glucoside).

A solution of silver trifluoromethanesulfonate (1.19 g, 35 4.64 mmol) in toluene (20 ml) was added during 5 minutes to a mixture of (22R)-16α,17α-butylidenedioxy-6α,9α-difluoro-11β,21-dihydroxy-4-pregnene-3,20-dione (1.09 g,

2.32 mmol), 2,3,4,6-tetra-O-benzoyl- $\alpha$ -D-glucopyranosyl bromide (2.30 g, 3.48 mmol) and powdered 4 $\text{\AA}$  molecular sieves (8.0 g) in dichloromethane (100 ml) at -20°C under nitrogen. The temperature was allowed to rise to -10°C  
5 during 1 h. Pyridine (3.0 ml) was added, and after additional 30 min stirring 0.5 M sodium thiosulfate (50 ml). The mixture was filtered through a layer of Celite. The organic phase was washed with water, 1 M sulfuric acid, water and saturated sodium hydrogen carbonate,  
10 dried over magnesium sulfate and concentrated.  
Chromatography (column: 50 x 4.0 cm, eluent:  
dichloromethane/ethyl acetate 9/1 by volume) gave  
amorphous (22R)-16 $\alpha$ ,17 $\alpha$ -butylidenedioxy-6 $\alpha$ ,9 $\alpha$ -difluoro-  
11 $\beta$ -hydroxy-4-pregnene-3,20-dione-21-yl 2',3',4',6'-  
15 tetra-O-benzoyl- $\beta$ -D-glucopyranoside (2.03 g, 83%).

HPLC-analysis showed 96.4% purity.

20  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  0.92 (t, H-25), 0.95 (s, H-18), 1.41 (m, H-24), 1.56 (s, H-19), 4.01 (m, H-5'), 4.39 (m, H-11), 4.55 (t, H-22), 4.89 (d, H-16), 5.25 (d,  $J_{1',2'} = 7.9$  Hz, H-1'), 5.29 (2 m, H-6), 5.54 (dd, H-2'), 5.75 (t, H-4'), 5.91 (t, H-3'), 6.15 (broad s, H-4).

25 MS showed an  $[\text{M}+\text{Na}]^+$ ion of m/z 1069. (The calculated nuclide mass sum is 1046.4.)

Sodium methoxide in methanol (4.0 ml, 0.5M) was added to a solution of this material (1.11 g, 1.06 mmol) in  
30 dichloromethane/methanol (50 ml, 1/3 by vol) at room temperature. After stirring overnight, the solution was neutralized with Dowex 50 ( $\text{H}^+$ ) resin, filtered and concentrated. Chromatography (column: 30x4.0 cm, eluent:  
dichloromethane/methanol 5/1 by vol) gave the title  
35 compound as an amorphous material (554 mg, 83%).

HPLC-analysis showed 97% purity.

5        $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ ):  $\delta$  0.96 (s, H-18), 0.99 (t, H-25), 1.51  
1.60 (s, H-19), 3.70 (m, H-6'a), 3.93 (broad d, H-6'b), 4.33 (m, H-11), 4.38 (d,  $J_{1',2'} = 7.6$  Hz, H-1'), 4.60 (d, H-21a), 4.72 (t, H-22), 4.89 (d, H-21b), 5.45 (2m, H-6), 6.05 (broad s, H-4).

MS showed an  $[\text{M}+\text{H}]^+$ ion of m/z 631, and an  $[\text{M}+\text{H}]^+$ ion of m/z 653. (The calculated nuclide mass sum is 630.3).

10      Example 2.

(22R)-16 $\alpha$ ,17 $\alpha$ -Butylidenedioxy-11 $\beta$ -hydroxypregna-1,4-diene-3,20-dione-21-yl  $\beta$ -D-glucopyranoside (Budesonide 22R-epimer  $\beta$ -D-glucoside).

15      Budesonide (1.00 g, 2.32 mmol) was reacted with 2,3,4,6-tetra-O-benzoyl- $\alpha$ -D-glucopyranosyl bromide (2.30 g, 3.48 mmol) analogously to that described in example 1.

Chromatography (column: 50x4.0 cm, eluent:

20      dichloromethane/ethyl acetate 7/1 by vol) gave amorphous (22RS)-16 $\alpha$ ,17 $\alpha$ -butylidenedioxy-11 $\beta$ -hydroxypregna-1,4-diene-3,20-dione-21-yl 2',3',4',6'-tetra-O-benzoyl- $\beta$ -D-glucopyranoside (1.96 g, 84%).

25      HPLC-analysis showed 98.8% purity.

30       $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ): 0.87 (t, H-(S)25), 0.90 (t, H-(R)25), 0.98 (s, H-(R)18), 1.02 (s, H-(S)18), 1.50 (s, H-(RS)19), 5.21 (d,  $J_{1',2'} = 7.8$  Hz, H-(S)1'), 5.23 (d,  $J_{1',2'} = 7.8$  Hz, H-(R)1'), 5.54 (dd, H-(R)2'), 5.56 (dd, H-(S)2'), 5.74 (t, H-(S)4'), 5.76 (t, H-(R)4'), 5.92 (t, H-(RS)3'), 6.03 (broad s, H-(RS)4), 6.29 (dd, H-(S)2), 6.31 (dd, H-(R)2).

35      MS showed an  $[\text{M}+\text{Na}]^+$ ion of m/z 1031. (The calculated nuclide mass sum is 1008.4)

This material (1.22 g, 1.21 mmol) was deacylated and purified analogously to that described in example 1. The 22R- and 22S-epimers of the obtained material (674 mg, 94%) were separated by semipreparative HPLC (Apex Prepsil 5 ODS 8  $\mu$ m, 25x2.25 cm) using acetonitrile/water 23/77 as eluent. This gave the title compound as an amorphous material (280 mg, 83%).

HPLC-analysis showed 98.5% purity.

10  $^1\text{H-NMR}$ (CD<sub>3</sub>OD):  $\delta$  0.96 (t, H-25), 0.99 (s, H-18), 1.46 (m, H-24), 1.53 (s, H-19), 3.69 (m, H-6'a), 3.93 (d, H-6'b), 4.37 (d, J<sub>1,2</sub>, 7.7 Hz, H-1'), 4.47 (m, H-11), 4.59 (d, H-21a), 4.67 (t, H-22), 4.86 (d, H-21b), 4.90 (d, H-16), 15 6.06 (broad s, H-4), 6.30 (dd, H-2), 7.50 (d, H-1).

MS showed an [M+Na]<sup>+</sup>ion of m/z 615, and an [M+H]<sup>+</sup>ion of m/z 593. (The calculated nuclide mass sum is 592.3).

20 Example 3.

Sodium [(22R)-16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-6 $\alpha$ ,9 $\alpha$ -difluoro-11 $\beta$ -hydroxy-4-pregnene-3,20-dione-21-yl  $\beta$ -D-glucopyranosid]uronate (GCS<sup>1</sup>IV 22R-epimer  $\beta$ -D-glucosiduronate).

A solution of silver trifluoromethanesulfonate (1.38 g, 5.38 mmol) in toluene (25 ml) was added during 15 min to a mixture of (22R)-16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-6 $\alpha$ ,9 $\alpha$ -difluoro-11 $\beta$ ,21-dihydroxy-4-pregnene-3,20-dione (1.20 g, 30 2.56 mmol), methyl (2,3,4-tri-O-benzoyl- $\alpha$ -D-glucopyranosyl bromide)uronate (2.39 g, 4.10 mmol) and powdered 4 $\text{\AA}$  molecular sieves (9.0 g) in dichloromethane/toluene (125 ml, 4/1 by vol) at -20°C 35 under nitrogen. The temperature was allowed to rise to 10°C during 2 h. Pyridine (5.0 ml) was added, followed by 0.5 M sodium thiosulfate (70 ml). The reaction mixture

was worked up as described in example 1. Chromatography (column: 50x4.0 cm, eluent: toluene/dichloromethane/ethyl acetate 40/20/15 by vol) gave amorphous methyl [(22R)-  
5      16 $\alpha$ ,17 $\alpha$ -butylidenedioxy-6 $\alpha$ ,9 $\alpha$ -difluoro-11 $\beta$ -hydroxy-4-  
pregnene-3,20-dione-21-yl 2',3',4'-tri-O-benzoyl- $\beta$ -D-glucopyranosid]uronate (1.59 g, 64%).

HPLC-analysis showed 97.7% purity.

10      $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  0.89 (s, H-18), 0.94 (t, H-25), 1.44 (m, H-24), 1.53 (s, H-19), 3.64 (s,  $\text{COOCH}_3$ ), 4.34 (d, H-5'), 4.44 (m, H-11), 4.54 (d, H-21a), 4.60 (t, H-22), 4.90 (d, H-16), 4.91 (d, H-21b), 5.25 (d,  $J_{1',2'} = 7.6$  Hz, H-1'), 5.28 (2 m, H-6), 5.58 (dd, H-2'), 5.67 (t, H-4'), 5.94 (t, H-3'), 6.15 (broad s, H-4).

MS showed an  $[\text{M}+\text{Na}]^+$  ion of m/z 993. (The calculated mass sum is 970.4.)

20     Litium hydroxide in water (9.1 ml, 1.0 M) was added to a solution of this material (1.38 g, 1.42 mmol) in tetrahydrofuran/water (65 ml, 3/1 by vol) at 0°C. The solution was allowed to attain room temperature, and after stirring for 24 h the solution was neutralized with  
25     acetic acid (1.0 ml) and concentrated. The residue was purified by semipreparative HPLC (Apex Prepsil ODS 8  $\mu\text{m}$ , 25x2.25 cm) using ethanol/40 mM aqueous triethylammonium acetate pH 5.0 33/67 as eluent. Fractions containing the desired substance were pooled, desalted on a C-18 column  
30     (10 g, Isolute; International Sorbent Technology, Hengoed, Mid Glamorgan, U.K.) using a stepwise water/methanol gradient, and converted into the sodium form by ion exchange on a column (4x2.5 cm) of Dowex 50Wx2 ( $\text{Na}^+$ -form). Lyophilization gave the title compound  
35     as an amorphous material (305 mg, 32%).

HPLC-analysis showed 97.3% purity.

<sup>1</sup>H-NMR(CD<sub>3</sub>OD): δ 0.95 (s, H-18), 0.99 (t, H-25), 1.51 (m, H-24), 1.60 (s, H-19), 4.35 (m, H-11), 4.44 (d, J<sub>1',2'</sub>=7.6 Hz, H-1'), 4.73 (t, H-22), 4.74 (d, H-21a), 5.45 (2m, H-6), 6.05 (broad s, H-4).

5

Example 4.

Sodium [(22R)-16α,17α-butylidenedioxy-11β-hydroxypregna-10,1,4-diene-3,20-dione-21-yl β-D-glucopyranosid]uronate (Budesonide 22R-epimer β-D-glucosiduronate).

A solution of silver trifluoromethanesulfonate (238 mg, 0.928 mmol) in toluene (4.0 ml) was added to a mixture of (22R)-16α,17α-butylidenedioxy-11β,21-dihydroxypregna-1,4-diene-3,20-dione (200 mg, 0.464 mmol), methyl (2,3,4-tri-O-benzoyl-α-D-glucopyranosyl bromide)uronate (406 mg, 0.696 mmol) and powdered 4Å molecular sieves (1.2 g) in dichloromethane (10 ml) at -50°C under nitrogen. The temperature was allowed to rise to 0°C during 2 h. Pyridine (600 μl) was added, followed by 0.5 M sodium thiosulfate (10 ml). The reaction mixture was worked up as described in example 1. Chromatography (column: 30x3.0 cm, eluent: dichloromethane/ethyl acetate 5/1 by vol) gave amorphous methyl [(22R)-16α,17α-butylidenedioxy-11β-hydroxypregna-1,4-diene-3,20-dione-21-yl 2',3',4'-tri-O-benzoyl-β-D-glucopyranosid]uronate (397 mg, 91%).

HPCL-analysis showed 99.0% purity.

30

<sup>1</sup>H-NMR(CDCl<sub>3</sub>): δ 0.92 (s, H-18), 0.92 (t, H-25), 1.40 (m, H-24), 3.67 (s, COOCH<sub>3</sub>), 4.33 (d, H-5'), 4.54 (m, H-11,21a,22), 4.87 (d, H-16), 4.87 (d, H-21b), 5.25 (d, J<sub>1',2'</sub>=7.3 Hz, H-1'), 5.57 (dd, H-2'), 5.70 (t, H-4'), 5.93 (t, H-3'), 6.03 (broad s, H-4) 6.30 (dd, H-2).

MS showed an  $[M+Na]^+$ ion of m/z 955. (The calculated nuclide mass sum is 932.4).

5      Lithium hydroxide in water (2.5 ml, 1.0 M) was added to a solution of this material (360 mg, 0.386 mmol) in tetrahydrofuran/water (18 ml, 3/1 by vol) at 0°C. The solution was allowed to attain room temperature, and after 22 h the solution was neutralized with acetic acid (290  $\mu$ l) and concentrated. Chromatography (column: 30 $\times$ 2.0 cm, eluent: ethyl acetate/acetic acid/methanol/water 16/3/3/2 by vol), followed by desalting, ion exchange and lyophilization as described in example 3 gave the title compound as an amorphous material (220 mg, 91%).

15     HPLC-analysis showed 98.2% purity.

16      $^1H$ -NMR( $CD_3OD$ ):  $\delta$  0.96 (s, H-25), 0.98 (s, H-18), 1.47 (m, H-24), 1.53 (s, H-19), 4.43 (d,  $J_{1',2}$ , 7.6 Hz, H-1') 4.48 (m, H-11), 4.68 (t, H-22), 4.71 (d, H-21a), 4.86 (d, H-21b), 4.89 (d, H-16), 6.05 (broad s, H-4), 6.30 (dd, H-2), 7.52 (d, H-1).

17     The following non-limitative examples illustrate pharmaceutical preparations suitable for the compounds of the invention.

Example 5. Tablet

18     Tablets are prepared by conventional compression methods  
30    with the following composition

	Budesonide 22R-epimer $\beta$ -D-glucoside, budesonide 22R-epimer $\beta$ -D-glucosiduronate, GCS <sup>1</sup> IV 22R-epimer $\beta$ -D-glucoside or GCS <sup>1</sup> IV 22R-epimer $\beta$ -D-glucosiduronate	5 mg
	Lactose	80 mg
5	Microcrystalline cellulose	20 mg
	Crosspovidone	5 mg
	Polyvinylpyrrolidone	5 mg
	Magnesium stearate	2 mg

10    Example 6. Enteric tablet

The tablet from Example 5 is coated with

	Eudragit L30D	3.7 mg
15	PEG 6000	0.4 mg
	Talc	0.9 mg

Example 7. Delayed release capsule

20    Budesonide 22R-epimer  $\beta$ -D-glucoside, budesonide 22R-epimer  $\beta$ -D-glucosiduronate, GCS<sup>1</sup> IV 22R-epimer  $\beta$ -D-glucoside or GCS<sup>1</sup> IV 22R-epimer  $\beta$ -D-glucosiduronate (7.1 g) is mixed with 300 g lactose, 128 g microcrystalline cellulose, 75 g crosslinked polyvinylpyrrolidone and 25 g polyvinylpyrrolidone. The mixture is granulated with water and the wet mass is extruded and spheronized giving cores with approximative size of 1 mm. The cores are dried and sieved. The cores are coated with a mixture of 25 255 g Eudragit NE30D, 77 g magnesium stearate and 250 g water in a fluid bed apparatus. Finally an enteric coating consisting of 11 g Eudragit L30D dispersion, 3 g triethylcitrate and 15 g talc is sprayed on the spheres. The pellets are dried in the fluid bed apparatus, sieved 30 and filled in hard gelatine capsules.

35

Example 8. Gut microflora controlled release capsule

Budesonide 22R-epimer  $\beta$ -D-glucoside or budesonide 22R-epimer  $\beta$ -D-glucosiduronate (6.6 g) is suspended in a  
5 solution of 1 g of hydroxypropylmethylcellulose in 50 ml of water. The mixture is sprayed on to 510 g sugar spheres in a fluid bed apparatus. Thereafter a mixture of 85 g guar gum, 30 g (solid content) Eudragit RL30D and 15 g talc in totally 900 g of a 1:1 mixture of water and  
10 isopropanol is sprayed on the spheres. Finally an enteric coating consisting of 100 g Eudragit L30D dispersion, 3 g triethylcitrate and 15 g talc is sprayed on the spheres. The pellets are dried in the fluid bed apparatus, sieved and filled into hard gelatine capsules.

15

Example 9. Gut microflora controlled release capsule

GCS<sup>1</sup> IV 22R-epimer  $\beta$ -D-glucoside or GCS<sup>1</sup>IV 22R-epimer  $\beta$ -D-glucosiduronate (6.8 g) is suspended in a mixture of 15 g locust bean gum, 5 g (solid content) Eudragit RL30D and 2 g talc in totally 220 g of a 1:1 mixture of water and isopropanol. This mixture is sprayed on to 510 g of sugar spheres in a fluid bed apparatus. Then a mixture of 80 g locust bean gum, 40 g (solid content) Eudragit RL30D and  
25 15 g talc in totally 900 g of a 1:1 mixture of water and isopropanol is sprayed on the spheres. Finally an enteric coating consisting of 100 g Eudragit L30D dispersion, 3 g triethylcitrate and 15 g talc is sprayed on the spheres. The pellets are dried in the fluid bed apparatus, sieved and filled in hard gelatine capsules.

Pharmacological testing

The anticolitic activity of the new prodrugs has been  
35 demonstrated in the colitis model described below. To judge that the prodrugs fulfil the intended profile and are broken down in the gut to active

glucocorticosteroids, the model has been designed so that the compounds have been administered orally and the anti-inflammatory activity judged in distal colon.

5    In vivo test model

Oxazolone-induced colitis in rats.

This is an IBD-model in rats, creating a T-cell dependent colitis after intra-rectal challenge of the hapten  
10 oxazolone in previously skin-sensitized animals. The inflammation starts with an acute stage that 24 hours after challenge shows infiltration of inflammatory cells, an increased colon-wet weight (edema), hyperaemia and slight ulcerations. After some days more chronic  
15 inflammation has developed with a persistent wet-weight increase and with a dominance of T-cells in the cell-infiltrate.

Experimental procedures

20 Dark Agouti rats were sensitized by painting 12 mg oxazolone (in 0.3 ml acetone/95% ethanol (1:4) on the skin on two consecutive days. Seven days after the second sensitization, the animals were challenged in the colon via a rectal injection of 6 mg oxazolone emulsified in  
25 200 µl of equal parts Orabase® and peanut-oil. After sacrificing the animals four days after challenge, the distal colon was weighed to obtain the wet-weight. The colitis was measured as edema (increase of wet-weight of distal colon over that of saline treated normals). Thymus  
30 weight was recorded as an unwanted systemic glucocorticoid activity.

Treatment

The glucocorticosteroid-glycosides were solved in a  
35 minute amount of ethanol and diluted with 0.9% NaCl. The animals received 30 or 300 nmol/kg bodyweight of the steroids orally (by gavage 10 ml/kg body weight) for

three days, starting the day after challenge. Control animals were treated with NaCl. The treatment groups included 6 animals.

## 5 Results

	GCS and dose (nmol/kg)	Colon edema (% of colitic controls)	Thymus weight (% of colitic controls)
	Budesonide 30	101 ± 15	96 ± 4
10	Budesonide 300	85 ± 19	54 ± 3
	Budesonide β-D-glucosiduronate 30	90 ± 17	77 ± 7
15	Budesonide β-D-glucosiduronate 300	58 ± 4	45 ± 6
	GCS <sup>1</sup> IV 22-R-epimer β-D-glucosiduronate 30	73 ± 12	86 ± 4
20	GCS <sup>1</sup> IV 22-R-epimer β-D-glucosiduronate 300	37 ± 8	36 ± 2
	GCS <sup>1</sup> V 22-R-epimer β-D-glucosiduronate 30	28 ± 6	84 ± 4
25	GCS <sup>1</sup> V 22-R-epimer β-D-glucosiduronate 300	0 ± 9	32 ± 3

### Conclusion:

The table shows that the new compounds according to the invention have a higher oral anticolitic potency and efficacy than the prior art compounds, budesonide and budesonide β-D-glucosiduronate. While the latter compounds at the dose 300 nmol/kg reduced the colonic edema by maximally about 40%, the two new compounds at the same dose inhibited the edema by about 65% or even fully blunted the edema. GCS<sup>1</sup> V 22-R-epimer β-D-glucosiduronate induced a much stronger inhibition at the dose 30 nmol/kg than for the two prior art compounds at 300 nmol/kg, showing that the new compound was more than 10 times more potent.

40

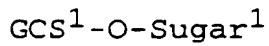
The new compounds also achieved a markedly better relation between antiedema efficacy and thymus

involution, which involution represents an unwanted systemic glucocorticoid activity. This is obvious when the three  $\beta$ -D-glucosiduronates are compared at the same dose levels: while the extent of thymus involution does 5 not differ so much, the colonic antiedema efficacy is much stronger for the new compounds.

Claims

1. A compound of the general formula

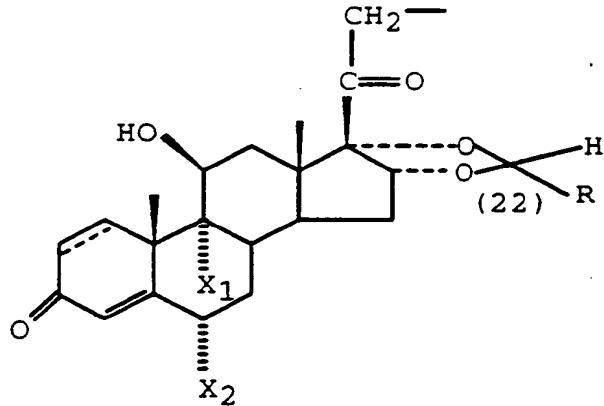
5



wherein  $\text{GCS}^1$  is a glucocorticosteroid ( $\text{GCS}^1\text{-OH}$ ) with high hepatic first pass metabolism, selected from the group of  
10 formula I

15

20

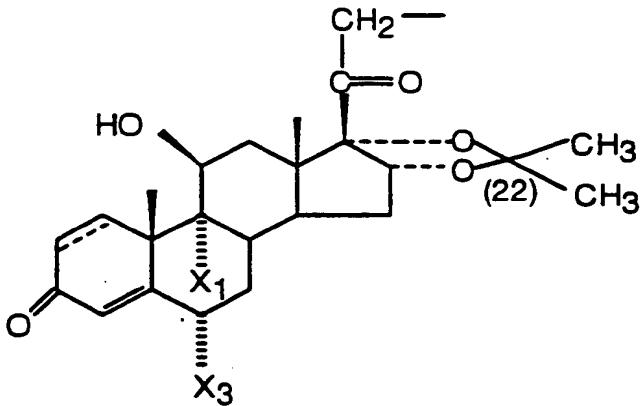


either as an epimeric mixture or as a corresponding pure 22R-epimer or 22S-epimer, or of the formula II

25

30

35



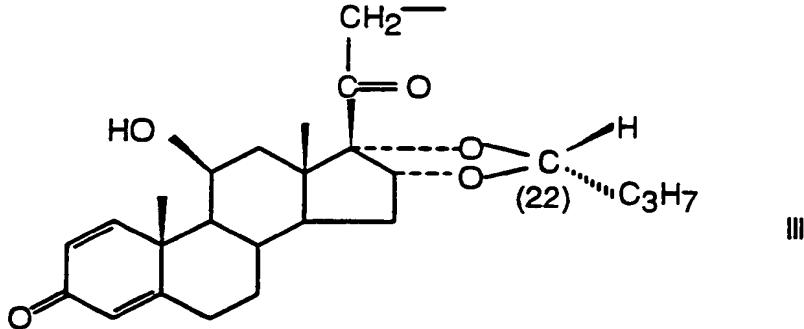
in which formulas  $X_1$  and  $X_2$  being the same or different hydrogen, fluoro, chloro or bromo substituents,  $X_3$  being a fluoro, chloro or bromo substituent, R being a hydrocarbon chain with 1-9 carbon atoms and in which  
5 formulas the 1,2-position is saturated or is a double bond, the Sugar<sup>1</sup> being the moiety of a monosaccharide, a disaccharide or an oligosaccharide, the GCS being linked in 21-position to the sugar via a glycosidic bond, as well as pharmaceutically and pharmacologically acceptable  
10 salts thereof.

2. A compound according to claim 1, wherein the GCS<sup>1</sup> is the 22R-epimer of formula I.

15 3. A compound according to claim 1, wherein the GCS<sup>1</sup> is the 22R-epimer of the budesonide moiety

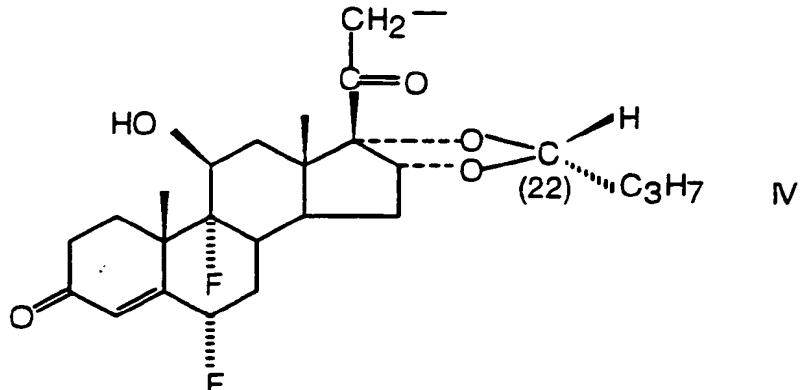
20

25



or the 22R-epimer of IV, with the formula

5

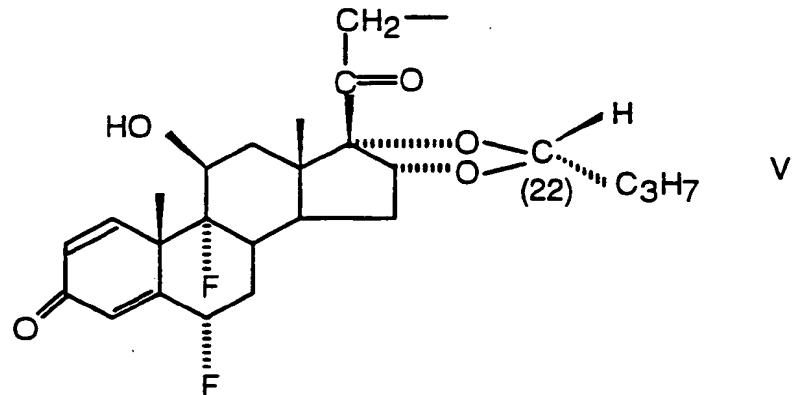


10

or the 22R-epimer of V, with the formula

15

20



25

4. A compound according to any of claims 1-3, wherein Sugar<sup>1</sup>-OH is D-glucose, D-galactose, D-cellulobiose or D-lactose.

30

5. A compound according to any of claims 1-3, wherein Sugar<sup>1</sup>-OH is D-glucuronic acid or D-galacturonic acid.

35

6. A compound according to claim 1, wherein Sugar<sup>1</sup>-OH is β-linked D-glucose.

7. A compound according to claim 1, wherein Sugar<sup>1</sup>-OH is  $\beta$ -linked D-glucuronic acid.

8. A compound according to claim 2, wherein Sugar<sup>1</sup>-OH  
5 is  $\beta$ -linked D-glucose.

9. A compound according to claim 2, wherein Sugar<sup>1</sup>-OH  
is  $\beta$ -linked D-glucuronic acid.

10 10. A compound according to claim 3, wherein Sugar<sup>1</sup>-OH  
is  $\beta$ -linked D-glucose.

15 11. A compound according to claim 3, wherein Sugar<sup>1</sup>-OH  
is  $\beta$ -linked D-glucuronic acid.

12. A compound according to claim 2, which is (22R)-  
16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-6 $\alpha$ -9 $\alpha$ -difluoro-11 $\beta$ -hydroxy-4-  
pregnene-3,20-dione-21-yl  $\beta$ -D-glucopyranoside.

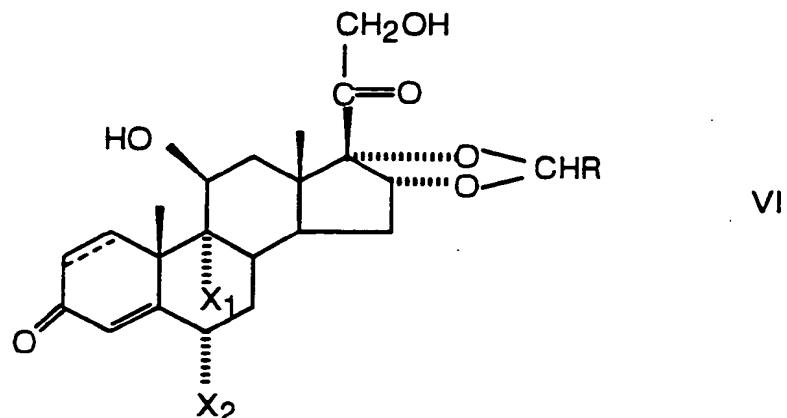
20 13. A compound according to claim 2, which is (22R)-  
16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-11 $\beta$ -hydroxypregna-1,4-diene-3,20  
dione-21-yl  $\beta$ -D-glucopyranoside.

14. A compound according to claim 2, which is  
25 sodium[(22R)-16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-6 $\alpha$ ,9 $\alpha$ -difluoro-11 $\beta$ -  
hydroxy-4-pregnene-3,20 dione-21-yl  $\beta$ -D-  
glucopyranosid]uronate.

15. A compound according to claim 2, which is  
30 sodium[(22R)-16 $\alpha$ ,17 $\alpha$ -butyldenedioxy-11 $\beta$ -hydroxypregna-  
1,4-diene-3,20 dione-21-yl  $\beta$ -D-glucopyranosid]uronate.

16. A process for the preparation of a compound  
according to claim 1, wherein the sugar is condensed with  
35 GCS<sup>1</sup>-OH of the formula VI, VI A, VI B or VII, in which  
 $X_1$ ,  $X_2$ ,  $X_3$ , R and the C<sub>1</sub>-C<sub>2</sub> bonding have the same meaning  
as defined in claim 1.

5



10

15

20

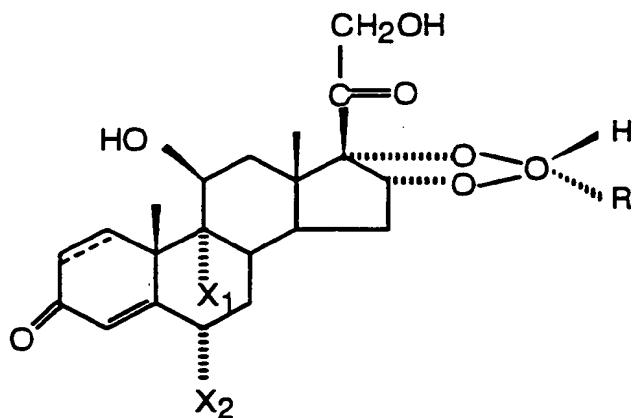
25

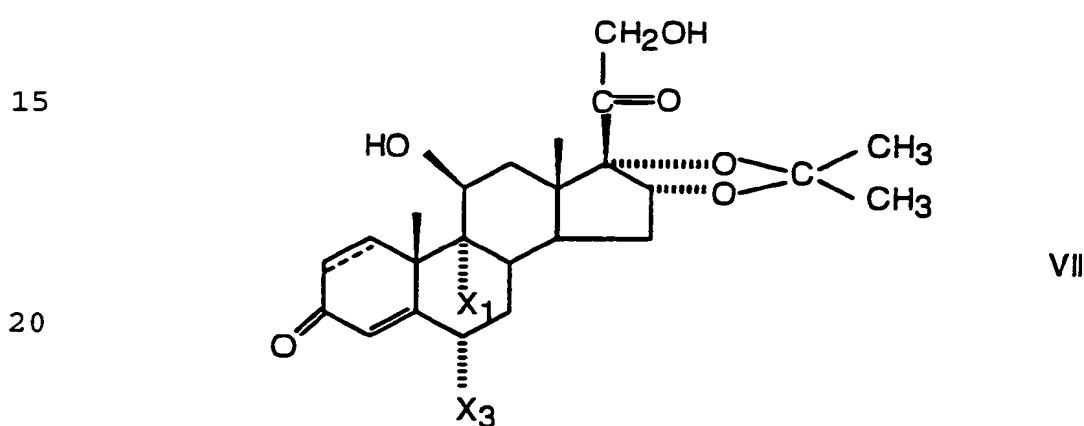
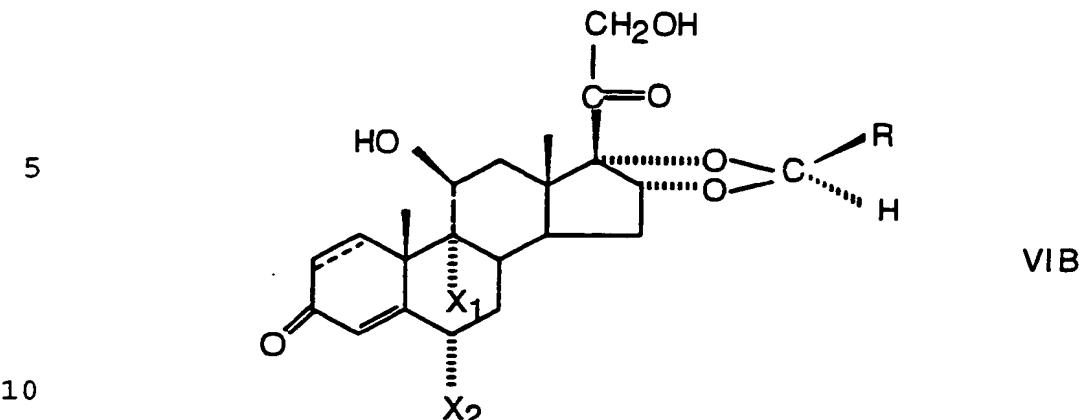
30

35

VI

VIA





25 17. A process according to claim 16 wherein a compound according to any of claims 2-15 is prepared.

18. A pharmaceutical formulation containing a compound according to any of claims 1-15 as an active ingredient.

30

19. A pharmaceutical formulation according to claim 18 in dosage unit form.

20. A pharmaceutical formulation according to any of  
35 claims 18-19 comprising the active ingredient in association with a pharmaceutical acceptable carrier.

21. A compound according to claim 1 for use in therapy.

22. The use of a compound according to claim 1 in the  
5 manufacture of a medicament for the treatment of inflamed bowel mucosa

23. The use of a compound according to claim 22 in the  
manufacture of a medicament for the treatment of  
10 ulcerative colitis.

24. The use of a compound according to claim 22 in the  
manufacture of a medicament for the treatment of Morbus  
Crohn.

## INTERNATIONAL SEARCH REPORT

1

International application No.

PCT/SE 93/01081

## A. CLASSIFICATION OF SUBJECT MATTER

IPC5: C07J 71/00, A61K 31/58

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC5: C07J

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE, DK, FI, NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

## CAS-ONLINE

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO, A1, 9107172 (AKTIEBOLAGET DRACO), 30 May 1991 (30.05.91), see the examples --	1-24
A	EP, A1, 0123485 (THE REGENTS OF THE UNIVERSITY OF CALIFORNIA), 31 October 1984 (31.10.84), see the whole document --	1-24
A	J. Med. Chem., Volume 27, No 3, 1984, David R. Friend et al., "A Colon-specific Drug-Delivery System Based on Drug Glycosides and the Glycosidases of Colonic Bacteria" page 261 - page 266 --	1-24

 Further documents are listed in the continuation of Box C. See patent family annex.

\* Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

- "X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

- "Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

- "Z" document member of the same patent family

Date of the actual completion of the international search

6 April 1994

Date of mailing of the international search report

14-04-1994

Name and mailing address of the ISA/  
Swedish Patent Office  
Box 5055, S-102 42 STOCKHOLM  
Facsimile No. +46 8 666 02 86Authorized officer  
**Göran Karlsson**  
Telephone No. +46 8 782 25 00

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/SE 93/01081

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	J. Med. Chem., Volume 28, No 1, 1985, David R. Friend et al., "Drug Glycosides: Potential Prodrugs for Colon-Specific Drug Delivery" page 51 - page 57  -- -----	1-24

**INTERNATIONAL SEARCH REPORT**  
Information on patent family members

26/02/94

International application No.	
PCT/SE 93/01081	

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
WO-A1- 9107172	30/05/91	AU-B-	643021	04/11/93
		AU-A-	7786491	13/06/91
		EP-A-	0502092	09/09/92
EP-A1- 0123485	31/10/84	JP-T-	60501105	18/07/85
		WO-A-	8404041	25/10/84